

PHYSICS 846

**Home Work Assignment # 5 (corrected version) 2/18/2009**

Due: Tues., Feb. 24, 2009.

1. *Dilute polyatomic gases:* Work through Sec. 6.1 of Kardar (p. 156 - 161).
2. Consider the lattice vibrations of a solid in  $d$  spatial dimensions. Further assume that the phonons in this solid have a dispersion  $\omega(k) = Ak^s$  with an arbitrary  $s > 0$ .
  - (a) Show that the low temperature specific heat then depends on temperature as  $C(T) = BT^a$  and determine the exponent  $a$  in terms of  $d$  and  $s$ .
  - (b) The experimentally measured specific heat of graphite over a certain range of temperature behaves as  $C \propto T^2$ . Suggest a plausible explanation for this behavior.
3. Starting with definition of the chemical potential as a derivative of the Helmholtz free energy:  $\mu(T, V, N) = (\partial F / \partial N)_{T, V}$ , show that it can also be written as a derivative of the entropy or internal energy.

$$\mu = -T \left( \frac{\partial S}{\partial N} \right)_{U, V} \quad \text{and} \quad \mu = \left( \frac{\partial U}{\partial N} \right)_{S, V}$$

4. The isothermal compressibility is defined as  $\kappa_T = -\frac{1}{V} (\partial V / \partial p)_{N, T}$ . Show that we can also write it in the very useful form

$$\kappa_T = \frac{1}{n^2} \left( \frac{\partial n}{\partial \mu} \right)_T$$

where  $n = N/V$  is the number density.

Hint: Argue, using the extensivity of the Helmholtz free energy, that  $F = Vf(n, T)$  where  $f$  is the free energy density, an intensive function of intensive variables. Then calculate  $\mu$ ,  $p$ ,  $1/\kappa_T$  etc. as derivatives of  $F$ .

5. (a) Show that the partition function of a classical ideal gas of  $N$  particles in the canonical ensemble at fixed  $T$  and  $V$  is given by

$$Z_N(N, T, V) = \frac{1}{N!} \left( \frac{V}{\lambda_T^3} \right)^N,$$

where  $\lambda_T$  is the thermal deBroglie wavelength.

(b) Using  $Z_N$  from part (a), show that the grand canonical partition function (or Gibbs sum) for the same gas at given  $\mu, T$  and  $V$  is given by

$$\mathcal{Z}(\mu, T, V) = \exp\left(\frac{yV}{\lambda_T^3}\right)$$

where  $y = \exp(\beta\mu)$  is called the *fugacity* (often denoted by  $z$ , but we'll call it  $y$  as there are too many  $z$ 's).

(c) Show that the probability there are  $N$  atoms in volume  $V$  of gas in diffusive contact with a reservoir is given by the Poisson distribution

$$P(N) = \alpha^N \exp(-\alpha) / N!$$

with  $\alpha = yV/\lambda_T^3$ . Check that this distribution is properly normalized and show that  $\alpha = \langle N \rangle$  the average number.

**6.** Prove that number fluctuations in the grand canonical ensemble are small in the thermodynamic limit. Toward this end, derive the following results, where the symbols have their standard meaning:  $\mathcal{Z}$  is the grand canonical partition function,  $\beta = 1/k_B T$ , and the partial derivatives with respect to the chemical potential  $\mu$  are taken with the temperature  $T$  and volume  $V$  held fixed. The angular brackets  $\langle \dots \rangle$  denote a grand canonical average.

(a)

$$\langle N \rangle = \frac{1}{\beta} \frac{\partial}{\partial \mu} \log \mathcal{Z}$$

(b)

$$\langle N^2 \rangle = \frac{1}{\beta^2} \frac{1}{\mathcal{Z}} \frac{\partial^2 \mathcal{Z}}{\partial \mu^2}$$

(c)

$$\langle (\delta N)^2 \rangle \equiv \langle (N - \langle N \rangle)^2 \rangle = \langle N^2 \rangle - \langle N \rangle^2$$

(d)

$$\langle (\delta N)^2 \rangle = \frac{1}{\beta} \frac{\partial \langle N \rangle}{\partial \mu}$$

(e)

$$\frac{\langle (\delta N)^2 \rangle}{\langle N \rangle^2} = \frac{n \kappa_T}{\beta N}$$

which in the limit of large volume vanishes (provided the compressibility is finite). What is the result of part (e) for the classical, ideal gas?