

Dr. Herbst

Answers To HW Set 2

1. Bernath p. 154, prob. 10 $E_n = -R_H c Z^2/n^2 \cong -4R/n^2$

$$R/hc \text{ (Bernath, p. 406)} = 109737 \text{ cm}^{-1} \quad \tilde{E}_n = -4\tilde{R}/n^2$$

$$\frac{1}{\lambda} = -4\tilde{R} \left(\frac{1}{n_u^2} - \frac{1}{n_l^2} \right) = 438948 \left(\frac{1}{n_l^2} - \frac{1}{n_u^2} \right)$$

$$\lambda(\text{\AA}) = 10^8 \lambda(\text{cm}) = 227.82 \left[\frac{1}{n_l^2} - \frac{1}{n_u^2} \right]^{-1}$$

Try $n_l = 1$ $n_u = 2, 3, 4, 5$ $\lambda(\text{\AA}) = 303.76, 256.29, 243.01, 237.31 \checkmark$

2. Bernath p. 154, prob. 11

(a) $\hat{O}(r)$: any radial operator $\langle \hat{O}(r) \rangle = \int_{r=0}^{\infty} |N_{n,l}|^2 R_{n,l}^* \hat{O} R_{n,l} r^2 dr$
 since spherical harmonics separately normalized over θ, ϕ .
 see Table 5.2 (p. 117)

$$R_{1,0}(r) = R_{1,0}(r) = \left(\frac{1}{a_0}\right)^{3/2} 2 \cdot e^{-r/a_0}$$

$$\langle \hat{O}(r) \rangle_{1s} = \frac{4}{a_0^3} \int_{r=0}^{\infty} e^{-2r/a_0} \hat{O}(r) r^2 dr$$

$$\int_0^{\infty} x^n e^{-ax} dx = n! / a^{n+1}$$

$$\therefore \langle r \rangle = \frac{4}{a_0^3} \int_{r=0}^{\infty} r^3 e^{-2r/a_0} dr = \frac{4}{a_0^3} \frac{3!}{(2/a_0)^4} = 1.5 a_0$$

$$\langle 1/r \rangle = \frac{4}{a_0^3} \int_{r=0}^{\infty} r e^{-2r/a_0} dr = \frac{4}{a_0^3} \frac{1!}{(2/a_0)^2} = \frac{1}{a_0}$$

(b) Table 5.3 $\psi_{1s} = \frac{1}{\sqrt{\pi}} \left(\frac{1}{a_0}\right)^{3/2} e^{-r/a_0}$

$$\psi_{2p_z} = \frac{1}{4\sqrt{2\pi}} \left(\frac{1}{a_0}\right)^{5/2} r e^{-r/2a_0} \cos\theta$$

$$\vec{\mu} = -e\vec{r} \quad \mu_x = -er \sin\theta \cos\phi \quad \mu_y = -er \sin\theta \sin\phi \quad \mu_z = -er \cos\theta$$

$$\left(\begin{matrix} \mu_x \\ \mu_y \\ \mu_z \end{matrix} \right)_{2p_z-1s} = -e \int_{r=0}^{\infty} \int_{\theta=0}^{\pi} \int_{\phi=0}^{2\pi} \psi_{2p_z}^* \begin{Bmatrix} r \sin\theta \cos\phi \\ r \sin\theta \sin\phi \\ r \cos\theta \end{Bmatrix} \psi_{1s} r^2 \sin\theta dr d\theta d\phi$$

$$\left(\begin{matrix} \mu_x \\ \mu_y \\ \mu_z \end{matrix} \right)_{2p_z-1s} = \frac{-e}{4\pi\sqrt{2} a_0^4} \int_{r=0}^{\infty} r^4 e^{-3r/2a_0} dr \int_{\theta=0}^{\pi} \begin{Bmatrix} \cos\theta \sin^2\theta \\ \cos\theta \sin^2\theta \\ \cos^2\theta \sin\theta \end{Bmatrix} d\theta$$

Since $\int_{\phi=0}^{2\pi} \cos\phi d\phi = \int_{\phi=0}^{2\pi} \sin\phi d\phi = 0$,

$$\times \int_{\phi=0}^{2\pi} \begin{Bmatrix} \cos\phi \\ \sin\phi \\ 1 \end{Bmatrix} d\phi$$

μ_z only non-zero component.

$$\therefore (\mu_z)_{2p_z-1s} = \frac{-e}{4\pi\sqrt{2} a_0^4} \underbrace{\int_{r=0}^{\infty} r^4 e^{-3r/2a_0} dr}_{\frac{4!}{(3/2a_0)^5}} \underbrace{\int_{\theta=0}^{\pi} \cos^2\theta \sin\theta d\theta}_{-\frac{1}{3} \cos^3\theta \Big|_0^{\pi}} \underbrace{\int_{\phi=0}^{\pi} d\phi}_{2\pi}$$

$\frac{2}{3}$

$$\therefore (\mu_z)_{2p_z-1s} = \frac{-ea_0}{4\pi\sqrt{2}} \frac{4 \times 3 \times 2}{3^5} 2^5 \frac{2}{3} 2\pi = \frac{-ea_0}{\sqrt{2}} \frac{2^8}{3^5}$$

$$(\mu_z)_{2p_z-1s} = -0.745 ea_0 \quad e = 4.803 \times 10^{-10} \text{ esu} \quad a_0 = 0.5292 \times 10^{-8} \text{ cm}$$

$$(\mu_z)_{2p_z-1s} = -1.893 \times 10^{-18} \text{ esu-cm} = -1.893 \text{ Debye}$$

3. Li (3 electrons) $Z=3$ $R = 13.6057 \text{ eV}$ (Bernath appendix)

$$E_{n_1 n_2 n_3} = -9R \left[\frac{1}{n_1^2} + \frac{1}{n_2^2} + \frac{1}{n_3^2} \right]$$

$1s^2 2s^1$ lowest configuration $E_{1,1,2} = -9R \left[\frac{1+1+1/4}{2.25} \right] = -275.52 \text{ eV}$

$$L=3 \quad S = 3/2, 1/2, 1/2 \Rightarrow {}^4F, {}^2F, {}^2F$$

$$(2) \quad L=2 \quad S = 3/2, 1/2, 1/2 \Rightarrow {}^4D, {}^2D, {}^2D \quad (2 \text{ each})$$

$$(3) \quad L=1 \quad S = 3/2, 1/2, 1/2 \Rightarrow {}^4P, {}^2P, {}^2P \quad (3 \text{ each})$$

$$L=0 \quad S = 3/2, 1/2, 1/2 \Rightarrow {}^4S, {}^2S, {}^2S$$

$$(b) \quad J = L+S, L+S-1, \dots |L-S| \quad {}^1S: J=0 \quad {}^3S: J=1 \quad {}^1D: J=2 \quad {}^3D: J=3, 2, 1$$

$${}^4F: J = 9/2, 7/2, 5/2, 3/2 \quad {}^2F: J = 7/2, 5/2$$

$${}^4D: J = 7/2, 5/2, 3/2, 1/2 \quad {}^2D: J = 5/2, 3/2$$

$${}^4P: J = 5/2, 3/2, 1/2 \quad {}^2P: J = 3/2, 1/2$$

$${}^4S: J = 3/2 \quad {}^2S: J = 1/2$$

$$5. \quad np^4 \quad W = \frac{6!}{2!4!} = 15 \quad (\text{same as } p^2) \text{ diagrams}$$

In class, terms for np^2 shown to be ${}^3P < {}^1D < {}^1S$. But, the 15 diagrams for np^2 and np^4 are related in the following sense:

m_L	1	$\uparrow\downarrow$	\leftrightarrow	$\begin{array}{c} \underline{00} \\ \underline{00} \\ \underline{\uparrow\downarrow} \end{array}$	or	$\begin{array}{c} \underline{\uparrow 0} \\ \underline{\uparrow\downarrow} \\ \underline{\uparrow 0} \end{array}$	\leftrightarrow	$\begin{array}{c} \underline{0\downarrow} \\ \underline{00} \\ \underline{0\downarrow} \end{array}$
	0	$\uparrow\downarrow$		$\underline{00}$		$\underline{\uparrow\downarrow}$		$\underline{00}$
	-1	$\underline{00}$	← hole	$\underline{\uparrow\downarrow}$		$\underline{\uparrow 0}$		$\underline{0\downarrow}$
M_L	2			-2	M_L	0		0
M_S	0			0	M_S	1		-1

For any np^4 diagram, construct an np^2 diagram such that their sum is np^6 ($M_L = M_S = 0$).

$$np^2 \Rightarrow M_L M_S \quad np^4 \Rightarrow -M_L, -M_S$$

Changing signs does not change L, S values enumerated.

$$\underline{nd^2} \quad W = \frac{10!}{2!8!} = 45 \quad \ell=2 \quad m_L = 2, 1, 0, -1, -2$$

	↑↓	↑	↑	↑	↑					↑	↑	↑	↑		↓	↓	↓	↓					
2	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—					
1	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—					
0	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—					
-1	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—					
-2	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—					
M_L	4	3	2	1	0	-1	-2	-3	-4	3	2	1	0	-1	-2	-3	3	2	1	0	-1	-2	-3
M_S	0	0	0	0	0	0	0	0	0	1	1	1	1	1	1	1	1	0	0	0	0	0	0

9 determinants $L=4$ $S=0$ 1G

	↓	↓	↓	↓					
2	—	—	—	—	—	—	—	—	—
1	—	—	—	—	—	—	—	—	—
0	—	—	—	—	—	—	—	—	—
-1	—	—	—	—	—	—	—	—	—
-2	—	—	—	—	—	—	—	—	—
M_L	3	2	1	0	-1	-2	-3		
M_S	-1	-1	-1	-1	-1	-1	-1		

$L=3$ $S=1$ 3F
21 determinants

↑↓	↑	↑			
—	—	—	—	—	
—	—	—	—	—	
—	—	—	—	—	
—	—	—	—	—	
—	—	—	—	—	
M_L	2	1	0	-1	-2
M_S	0	0	0	0	0

$L=2$ $S=0$ 1D
5 determinants

↑	↑	↓	↓	↓	↓		
—	—	—	—	—	—		
—	—	—	—	—	—		
—	—	—	—	—	—		
—	—	—	—	—	—		
—	—	—	—	—	—		
M_L	1	0	-1	0	-1	0	-1
M_S	1	1	0	0	0	-1	-1

$L=1$ $S=1$ 3P
9 determinants

m_L	2	1	0	-1	-2
M_S	0	0	0	0	0

$L=S=0$ 1S
1 determinant

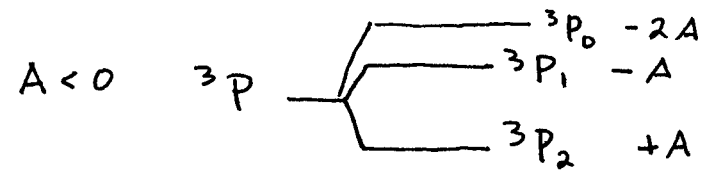
Order: $^3F < ^3P < ^1G < ^1D < ^1S$

p^4 Spin-Orbit Splitting

$^1S \rightarrow ^1S_0$ $^1D \rightarrow ^1D_2$ $^3P \rightarrow ^3P_0, ^3P_1, ^3P_2$

For 3P $k=1$ $S=1$ $J=2, 1, 0$

$$E_{fs}^{(1)} = \frac{A}{2} [J(J+1) - L(L+1) - S(S+1)] = \frac{A}{2} [J(J+1) - 4]$$



6. $N(1s^2 2s^2 2p^3)$ ground configuration $W = \frac{6!}{3!3!} = 20$

$m_L=0$ $M_S=3/2 \Rightarrow ^4S$ ground term

Two possible first excited configurations: $1s^2 2s 2p^4$
 $1s^2 2s^2 2p^2 3s$

Terms For $1s^2 \underbrace{2s 2p^4}_{sp^4}$ = Terms For $1s^2 2s^2 \underbrace{2p^2 3s}_{sp^2}$

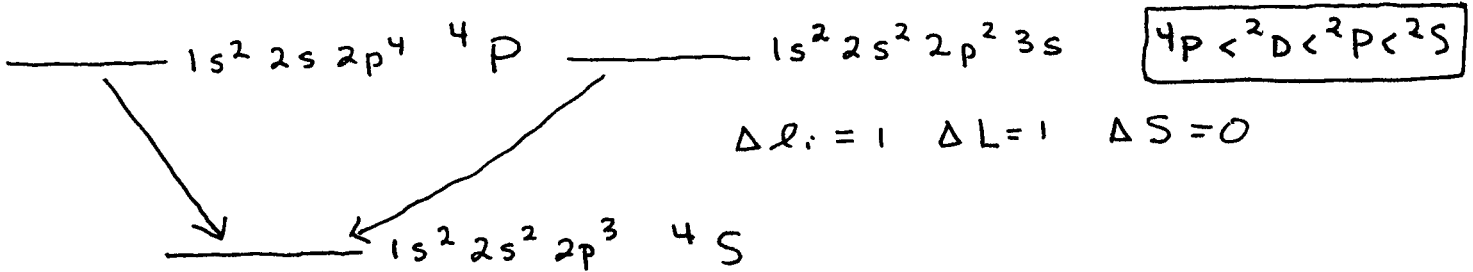
Terms For $p^2 = p^4$ $3P (L'=S'=1)$ $1D (L'=2 S'=0)$
 $1S (L'=S'=0)$

Now add s electron vectorially
 $L''=0$ $S''=1/2$

$3P (L'=1 S'=1) + S''=1/2 \Rightarrow L=1 S=3/2, 1/2$ $4P^2 P$

$1D (L'=2 S'=0) + S''=1/2 \Rightarrow L=2 S=1/2$ $2D$

$1S (L'=0 S'=0) + S''=1/2 \Rightarrow L=0 S=1/2$ $2S$



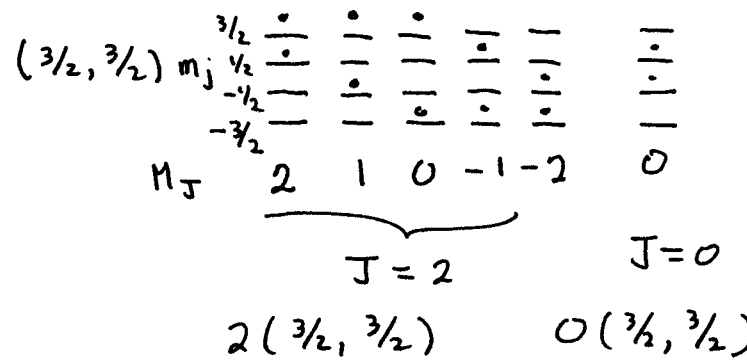
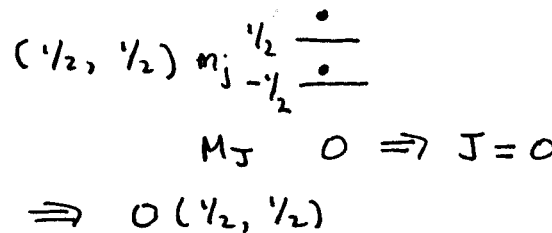
7. np^2 jj coupling For each electron $l=1$ $s=1/2$ $j=3/2, 1/2$

$(3/2, 1/2) \Rightarrow J=2, 1$ vector addition allowed

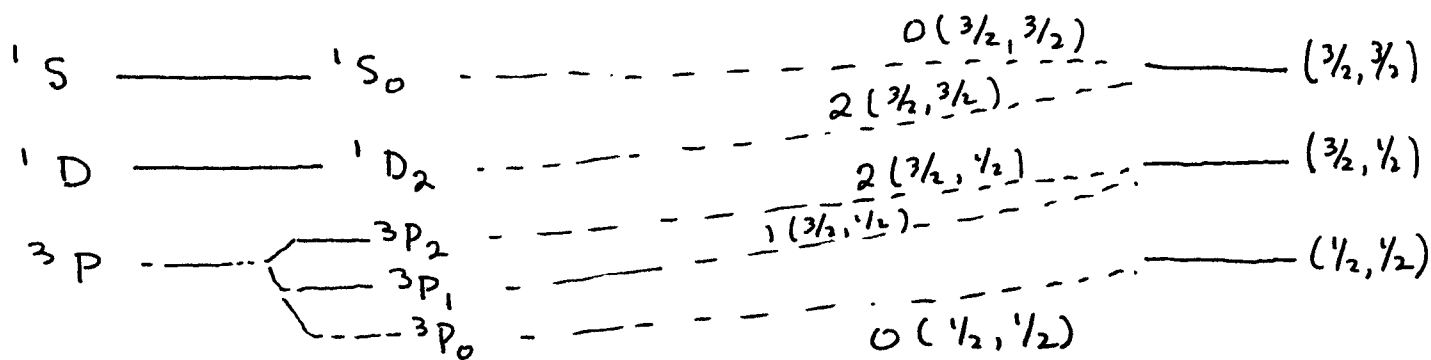
Note: no distinction between $(3/2, 1/2)$ + $(1/2, 3/2)$

states: $2(3/2, 1/2)$ and $1(3/2, 1/2)$

Now consider $(1/2, 1/2)$ + $(3/2, 3/2)$ diagrammatically:



Correlation Diagram



8. $E_{M_L M_S}^{(1)} = \mu_b B_z (M_L + 2M_S) + A M_L M_S$ (in class)

$2P \quad L=1 \quad S=1/2 \quad M_L = 1, 0, -1 \quad M_S = \pm 1/2$

M_L	M_S	$E_{M_L M_S}^{(1)}$
1	$1/2$	$2\mu_b B_z + A/2$
0	$1/2$	$\mu_b B_z$
-1	$1/2$	$-A/2$
1	$-1/2$	$-A/2$
0	$-1/2$	$-\mu_b B_z$
-1	$-1/2$	$-2\mu_b B_z + A/2$

