

# Preview of Period 2: Electromagnetic Waves - Radiant Energy I

## 2.1 Energy Transmitted by Waves

How can waves transmit energy?

## 2.2 Refraction of Radiant Energy

What happens when a light beam travels through water?

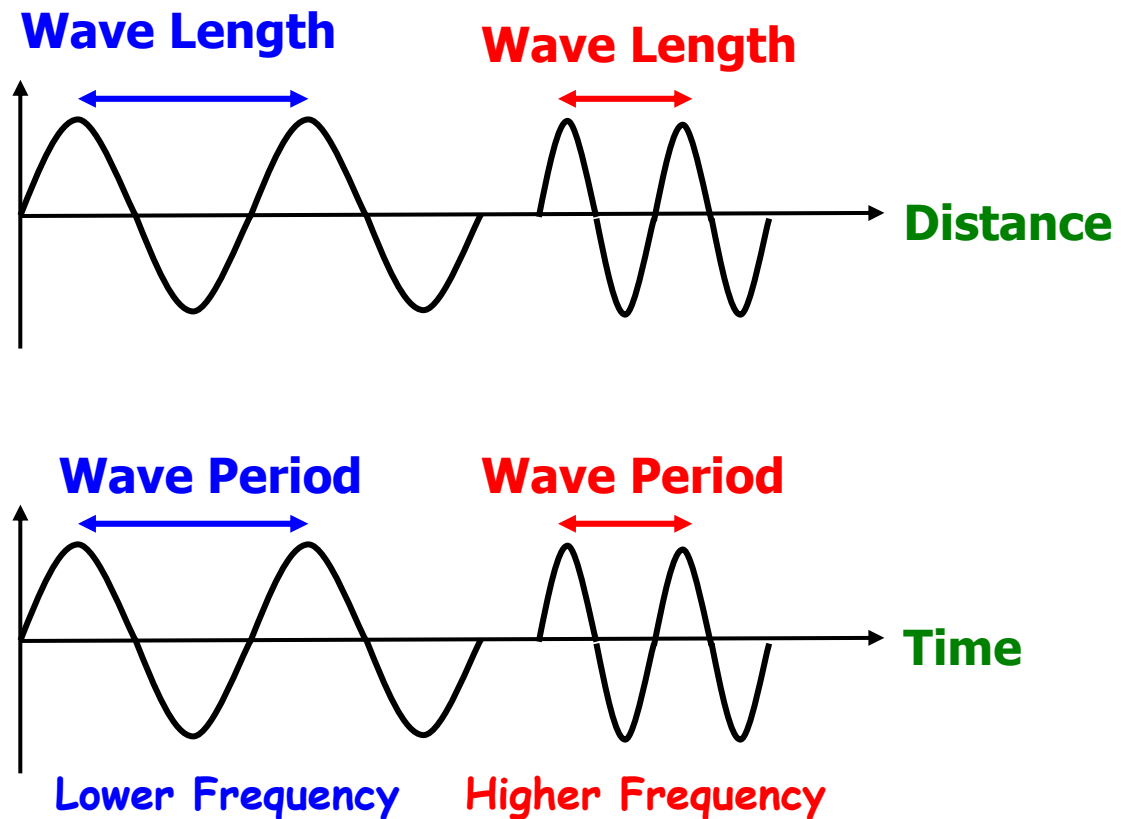
## 2.3 Focusing Radiant Energy

What happens when a light beam strikes a lens?

## 2.4 The Quantum Model of Electromagnetic Radiation

How can the quantum model explain the electric current produced by solar cells and fluorescence?

## Wavelength, Period, and Frequency



The **period** of a wave is the time it takes the wave to complete one cycle.

The **frequency** of a wave is the inverse of its period.

$$\text{frequency} = 1/\text{period}$$

Frequency is measured in Hertz (Hz).

$$1 \text{ Hz} = 1 \text{ cycle/second}$$

## Wave Speed

The **frequency** of a wave is the inverse of its period. Frequency is measured in Hertz (Hz).

$$\text{frequency} = 1/\text{period}$$

The relationship between **wavelength** and **frequency** gives the **speed** of a wave:

$$s = fL$$

**s** = speed at which radiant energy travels  
(meters/sec or mi/sec)

**f** = frequency (cycles/sec, or Hertz)

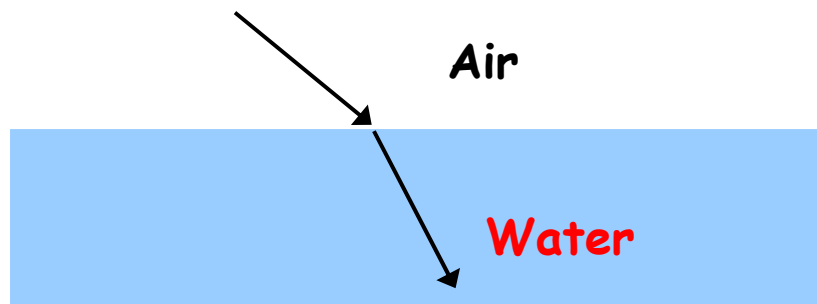
**L** = wavelength (in meters or feet)

### Compare Electromagnetic and Sound Waves

- ◆ Which type of waves can travel through **air**?
- ◆ Which type of waves can travel through a **vacuum**?
- ◆ How do the **speeds** of sound waves and electromagnetic waves compare?

## Refraction of Light

When light enters a transparent material, the speed of the wave changes and the light beam is *refracted*, or bent.



The *index of refraction* ( $n$ ) is a measure of the amount that a light beam is bent as it passes from one medium to another medium.

$$n = \frac{\text{speed of light in a vacuum}}{\text{speed of light in material}}$$

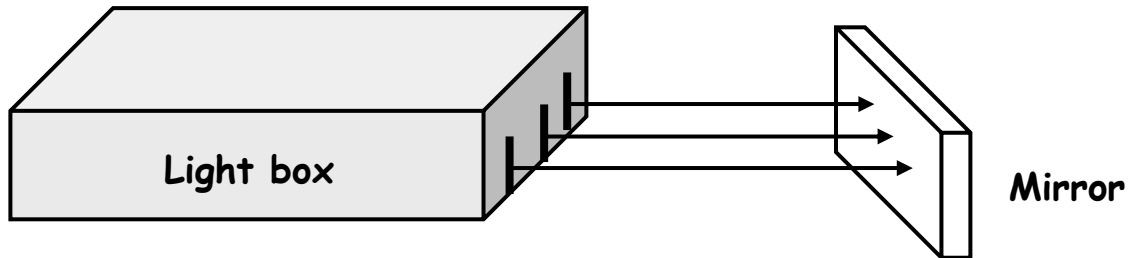
The speed of light in a vacuum is  $3 \times 10^8$  m/s.

## Light Refracted by a Prism

The amount that light is refracted depends on the frequency of the light wave.

When light passes through a prism, this difference in refraction separates the light into a rainbow of colors.

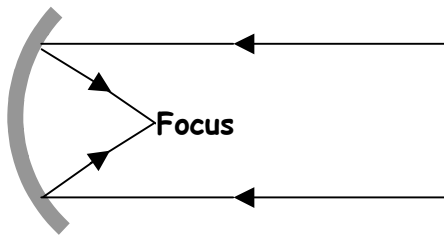
## Light Reflecting from a Plane Mirror



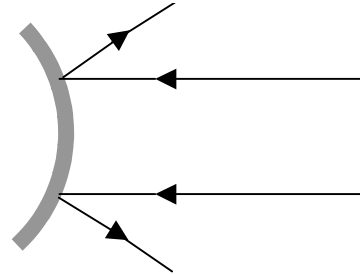
- ◆ Place a sheet of white paper under the mirror.
- ◆ Mark the position of the mirror on the paper with a straight line.
- ◆ Draw lines on the paper showing the path of the light beams as they strike the mirror and reflect off the mirror.
- ◆ Use a protractor to compare the **angle of incidence** and the **angle of reflection** of light striking a mirror.
- ◆ How are these angles related?

# Mirrors and Lenses

## Light from concave and convex mirrors

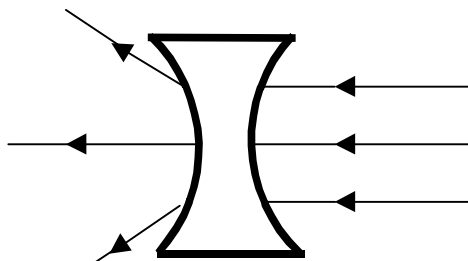


**A Concave Mirror Focuses  
Radiant Energy**

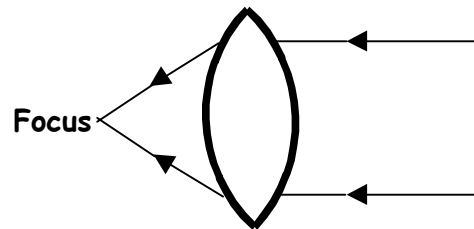


**A Convex Mirror Spreads  
Radiant Energy**

## Light through concave and convex lenses



**A Concave Lens Spreads  
Radiant Energy**



**A Convex Lens Focuses  
Radiant Energy**

# Quantum Model of Electromagnetic Radiation

The quantum model treats radiant energy as many **small packets of energy** called **photons**.

The energy of a photon is related to the frequency and wavelength.

$$E = h f = (h c)/L$$

***E*** = energy of a photon (joules)

***h*** = is a proportionality constant  
=  $6.63 \times 10^{-34}$  joule sec

***f*** = frequency (Hertz)

***c*** = speed of the radiant energy  
=  $3 \times 10^8$  meters/sec

***L*** = wavelength (meters).

When radiant energy interacts with matter, it absorbs or deposits energy in amounts that are integer multiples of this photon energy

## Solar Cells and the Quantum Model

The quantum model treats radiant energy as many **small packets of energy** called **photons**.

Solar cells use the **photoelectric effect** to produce electricity.

- ◆ When electrons absorb photons of electromagnetic radiation, some electrons have enough energy to escape from their atom and form an electric current.
- ◆ Only photons with wavelengths equal to or shorter than visible light have enough energy per photon to produce a current.

## Period 2 Summary

- 2.1** The period of a wave is the time it takes to complete one cycle. The frequency of a wave is the inverse of its period: **frequency = 1 / period**

Radiant energy can be thought of as a wave with a wavelength and a frequency. The speed of a wave is

$$s = f L$$

- 2.2** As light passes from one medium to another it is refracted, or bent. Light travels at  $3.0 \times 10^8$  m/s in a vacuum, but travels at different speeds in materials such as in water or glass. The index of refraction is

$$n = \frac{\text{speed of light in a vacuum}}{\text{speed of light in a material}}$$

- 2.3** A concave mirror focuses beams of radiant energy, but a convex mirror spreads the beams.

A convex lens focuses radiant energy and a concave lens spreads the energy.

- 2.4** The quantum model treats radiant energy as consisting of small packets of energy called photons. Photon energy is related to wave frequency or wavelength.

$$E = hf = (hc)/L$$

- 2.5** In solar cells (photoelectric cells), electrons can absorb photons of radiant energy. A photon of sufficient energy can cause an electron to escape from its atom and generate an electric current.

## Period 2 Review Questions

- R.1** What is the difference between wavelength, wave amplitude, and wave frequency? Which of these variables can be used to determine the speed of a wave?
- R.2** Compare the speed of sound to the speed of light in air. What is the ratio of the speed of sound to the speed of light?
- R.3** What is refraction of light? Why is white light that travels through a prism split into its constituent colors?
- R.4** Why is it unsafe to leave glass soft drink bottles in a forest?
- R.5** Photons striking a particular solar cell do not produce an electric current in the cell. Why is this? Would more photons of the same energy produce a current?